

# MINISTRY OF EDUCATION, CULTURE, RESEARCH, AND TECHNOLOGY UNIVERSITAS NEGERI JAKARTA FACULTY OF MATHEMATICS AND NATURAL SCIENCE CHEMISTRY STUDY PROGRAM

Kampus A, Gedung Hasjim Asj'arie Rawamangun, Jakarta Timur 13220 Telp/Fax : (021) 4894909, E-mail : pkimia@unj.ac.id

## **Bachelor in Chemistry**

### **Module Handbook**

Module name:	Basic Chemistry Practicum							
Module level, if applicable:	Undergraduate							
Code:	33250112							
Sub-heading, if applicable:	-							
Classes, if applicable:	-							
Semester:	$2^{nd}$							
Module coordinator:	Dr. Fera Kurniadewi, M.Si.							
Lecturer(s):	Dr. Hanhan Dianhar, M.Si. Edith Allanas, M.Pd.							
Language:	Bahasa Indonesia (Indonesian Language)							
Classification within the curriculum:	Compulsory Courses in the first year (2 <sup>nd</sup> semester) Bachelor Degree							
Teaching format/class hours per week during the semester	Learning activity can be carried out in the form of Laboratory activity: 340 minutes per week  - Safety induction: 1 time (MSDS, safety equipment, waste disposal)  - Preparation: 1 time (chemical preparation and experiment equipment)  - Laboratory work: 11 times (8 project topics, i.e pretest, practicum activity, and writing report)  Discussion: 340 minutes for 1 time (presentation and discussion of practical results)  Examination: 340 minutes for 2 times (mid and final examination)							
Workload:	Type	CU	Laboratory Activity	Discussion	Examination			
	Р	2	73,66 h 2,44 ECTS	5,66 h 0,188 ECTS	11,33 h 0,372 ECTS			
Credit points:	2 CU (3 ECTS)							
Prerequisite course(s):	Basic Chemistry 1 and Basic Chemistry 2							
Course outcomes:	CLO1. Demonstrate basic skills and work safety in chemical laboratories CLO2. Carry out chemical experiments according to practical guidelines and analyze the data obtained							



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Content:	<ol> <li>Basic OHS Skills in Chemistry Lab</li> <li>Basic Skills in Chemistry Laboratory</li> <li>Basic laws of chemistry</li> </ol>						
	<ul> <li>4. Chemical reaction</li> <li>5. Thermochemistry</li> <li>6. Reaction rate</li> <li>7. Chemical equilibrium</li> <li>8. Acid-base and acid-base indicators</li> <li>9. Colligative Properties of Solutions</li> <li>10.Colloid</li> <li>11.Redox and Electrochemistry</li> </ul>						
Study/exam achievements:	Examinations are conducted as Unit Tests. There are two-unit tests, each covers 4-5 chapters. The final marks are derived from unit tests (70%) and structured tasks (30%).  Aspect (%) Attitude 15						
	General skills Special skills Knowledge Final score	10 50 25 100					
Media	Laboratory equipment, Projectors, Practical videos, Learning Management System (MsTeams or Alkana)						
Literatures	<ol> <li>Tim, 2021, Practical Guide Basic Chemistry Practicum Departement Chemistry FMIPA Universitas Negeri Jakarta</li> <li>Elias, A. J., A., Collection of Interesting General Chemistry Experiments, Sangam Books, 2002.</li> <li>Murov, S. L., Experiments in General Chemistry (Cengage Laboratory Series for General Chemistry), Cengage Learning, 2014.</li> </ol>						

## PLO and CO mapping

	PLO1	PLO2	PLO3	PLO4	PLO5	PLO6	PLO7	PLO8	PLO9	PLO10	PLO11	PLO12
CO1	v		v					v				
CO2	v		v					v				